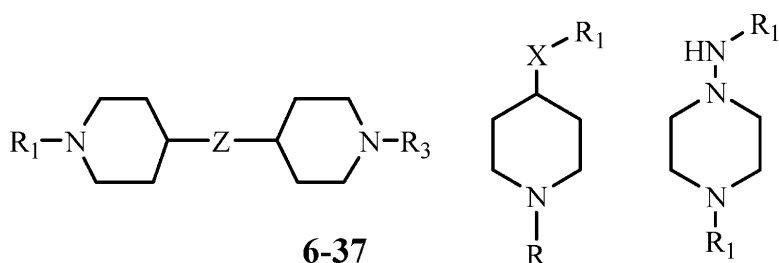


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Design, Synthesis, and Preliminary Pharmacological Evaluation of a Set of Small Molecules That Directly Activate Gi Proteins

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Heterotrimeric G proteins play a pivotal role in the communication of cells with the environment. G proteins are stimulated by cell surface receptors (GPCR) that catalyze the exchange of GDP, bound to G α subunit, with GTP and can per se be the target of drugs. Based on the structure of two nonpeptidic modulators of Gi proteins, a series of new molecules characterized by a long hydrophobic chain and at least two nitrogen atoms protonated at physiological pH was designed. The compounds were tested for their ability to stimulate binding of GTP γ S to recombinant Gi proteins. Gi activation properties were also evaluated by inhibition of adenyl cyclase activity in intact lymphocytes. Most compounds were able to stimulate GTP γ S binding and to inhibit cAMP production at micromolar doses. Among the active compounds, **34** showed good efficacy and was the most potent compound studied, particularly on α_o subtype; its regioisomer, **36**, was the most efficacious one. Compound **7** showed also an interesting profile as it showed selectivity toward the α_o subtype, in both efficacy and potency. Some of the compounds synthesized and found to be active may be useful leads to develop more potent and selective Gi protein modulators.

Introduction

Heterotrimeric G proteins play a pivotal role in the communication of cells with the environment. G proteins are stimulated by cell surface receptors [G protein-coupled receptors (GPCR)] that catalyze the exchange of GDP, bound to G α subunit, with GTP. Consequently, the GTP-bound G protein dissociates into two signaling entities: the G α subunit and the G $\beta\gamma$ subunit complex; β and γ subunits form tight complexes that cannot be dissociated under non-denaturing conditions.¹

G α and G $\beta\gamma$ both modulate cellular effectors. Today, several subtypes of G α subunits are known, and stimulatory or inhibitory families can be identified. Inactivation occurs by the intrinsic GTPase activity of the G α subunit; GDP-bound G α subunit reassociates with G $\beta\gamma$ subunits, and this causes mutual inactivation because the effector's binding surfaces are inaccessible in the oligomer.

In this cycle of activation and deactivation, specific binding sites on G proteins allow for the sequential, conformation-dependent binding of protein reaction partners. These include receptors that interact with all three subunits, effectors, and regulators of G protein signaling (RGS), which bind to the transition state of G α -GTP and exert a GTPase activating effect. The increased GTP turnover not only accelerates the rate

of signal deactivation but also enhances the rate of activation.^{2,3} In addition, several modulators have been identified that activate G protein signaling in the absence of a GPCR and that cannot be readily placed into this basal reaction cycle. These include activators of G protein signaling such as AGS1 peptide that apparently promotes nucleotide exchange on Gi proteins,⁴ PCP2, a nucleotide exchange factor for G α_o ,⁵ AGS2 that, in protein interaction studies, selectively associates with G $\beta\gamma$,^{6,7} and AGS3⁷ that interacts with G α -GDP, preventing the interaction of G α -GDP with G $\beta\gamma$; alternatively, it has been proposed that AGS3 may simply compete with G $\beta\gamma$ for binding to the G α subunit.⁸ Thus, several specific binding sites exist on the G α subunit that may be exploited for the design of synthetic stimulatory or inhibitory compounds. In both experimental pharmacology and clinical pharmacotherapy, G protein-dependent signaling pathways are activated or inhibited by employing appropriate receptor agonists or antagonists, respectively. Several arguments indicate that G proteins can per se also be targeted by drugs and that this approach may, at least conceptually, offer advantages.¹ One of the most convincing arguments in favor of direct G protein targeting is the known promiscuity of receptors that often are able to couple with different classes of G proteins.⁹ Thus, cellular stimulation by a receptor may result in the concerted activation of several distinct G proteins driving the activation of multiple effectors pathways. In principle, it might be useful to block signaling of a receptor via one type of G protein but not via the other type, a goal that cannot

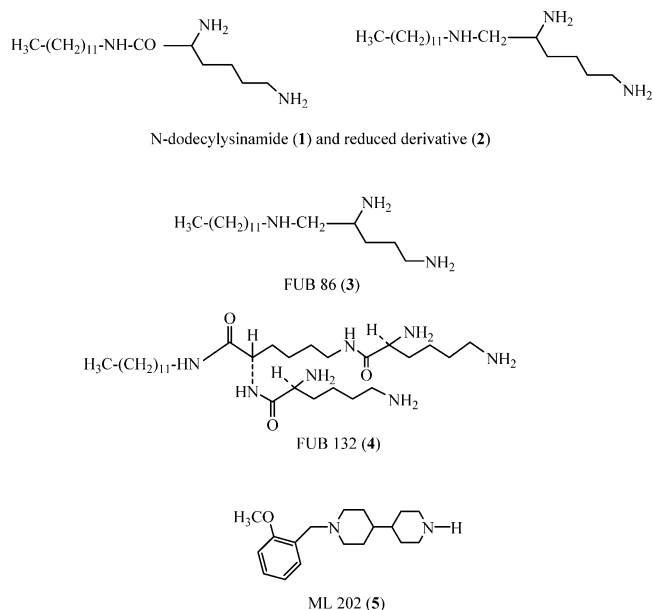
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Chart 1

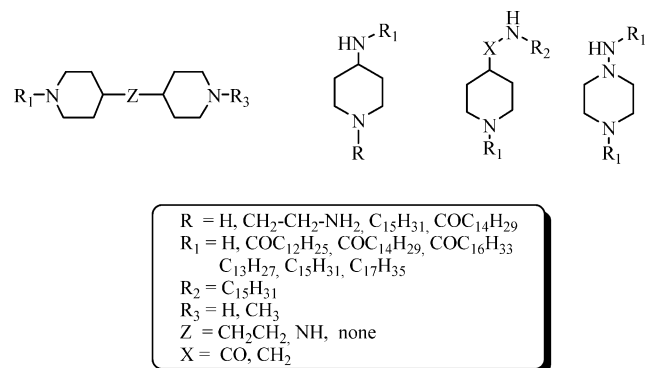


be achieved with receptor antagonists but that may be possible with direct interaction with the individual G protein.¹⁰ Some low molecular weight compounds have been identified that bind directly to G proteins.^{11–13} Suramin and analogues such as NF 023 have been reported to act as nonselective G protein inactivators.^{10,14,15} Mechanistic aspects have been studied in detail for receptor-derived peptides¹⁶ and the receptor-mimetic peptide Mastoparan.^{17,18}

Mastoparan (a 14 aa peptide from wasp venom), by mimicking the structure of activated receptor, selectively stimulates inhibitory G α subunits (G $\alpha_{i/o}$). On the other hand, nonpeptidic modulators of G proteins are rare. As a matter of fact, because these molecules have to penetrate the membrane, their pharmacokinetic properties must be quite stringent. Some amphiphilic β -adrenoreceptor antagonists such as propranolol and local anesthetics such as bupivacain have been reported to be direct activators of G α proteins, but at fairly high doses,¹⁹ and the same has been claimed about some taste substances such as quinine and aspartame.²⁰ In 1997 Leschke and co-workers²¹ described alkyl-substituted amino acid amides and analogues that act as new synthetic nonpeptidic direct G $\alpha_{i/o}$ protein activators such as *N*-dodecyl lysineamide (1), its reduced derivative (2), and FUB 86 (3) (Chart 1); later, a more efficient compound, FUB 132 (4), was developed.²² More recently, Melchiorre and co-workers²³ reported the protein G α activating properties of several molecules including compound 5 (Chart 1).

Nearly all inhibitory neurotransmitters able to enhance the pain threshold utilize G α proteins as signal transduction system. G α proteins represent the most widespread modulatory signaling pathway in neurons,²⁴ and their involvement in the modulation of pain perception has been well established. The administration of pertussis toxin (PTX), which selectively inactivates G α proteins, produced hyperalgesia and allodynia in laboratory animals.^{25,26} An increase of pain perception has also been observed in knock-out mice that lack the gene coding the subtype G α_o of the G α protein family.²⁷

Chart 2



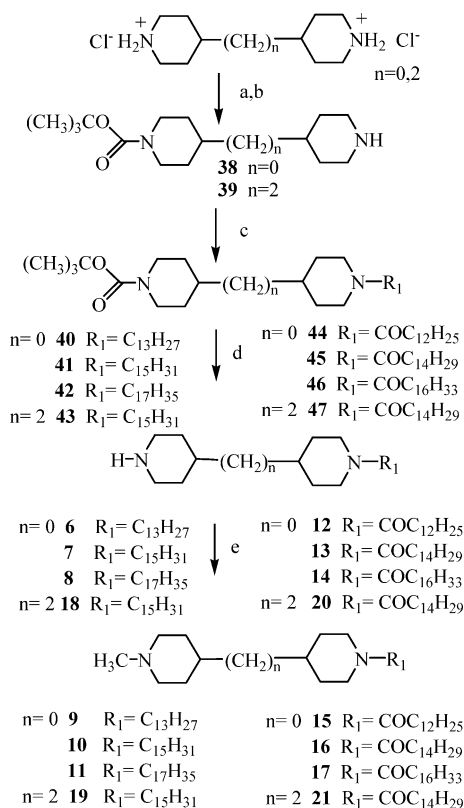
These data clearly indicate that a lack of functionality of G α proteins enhances the sensitivity to pain.

Hypofunctionality of G α proteins also produced insensitivity to analgesic treatments acting via G α stimulation. Thus, it has been observed that PTX prevents the enhancement of pain threshold induced by a wide variety of drugs with analgesic properties such as opioids, α_2 -agonists, tricyclic antidepressants, and antihistamines.^{25,28,29} These results are also supported by antisense studies in which the inhibition of the expression of the G α_i protein subunits prevented the analgesia induced by antihistaminic drugs and tricyclic antidepressants.^{30,31} A selective involvement of specific G α protein subtypes in the mechanism of action of analgesic compounds has also been demonstrated. The administration of selective antibodies against G $\alpha_{i/2}$, as well as the inhibition of its expression by the use of specific antisense oligonucleotides, prevented the analgesia induced by agonists of μ -opioid receptors.^{32,33} The use of an antisense oligonucleotide (aODN) against G $\alpha_{i/3}$ demonstrated the essential role played by the G $\alpha_{i/3}$ subtype in increasing the pain threshold induced by α_2 -adrenoceptor agonists.³⁴

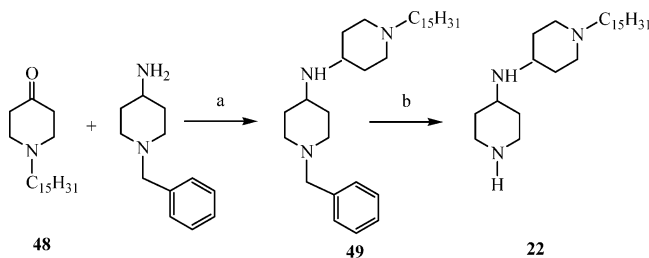
In 2001, some of us reported that hypofunctionality of G α proteins is the etiopathogenic mechanism for migraine and cluster headaches.³⁵ This discovery has of course raised interest in small molecules able to restore G α functionality, bypassing the receptors implied, to treat this pathological state, and we have started research to identify molecules directly activating G α proteins. In this respect it is interesting that most major analgesics (morphine, baclofen, clonidine) are known to act through receptors that induce G α protein activation.²⁵ Thus, in theory, identification of selective and potent G α activators could result in the discovery of potent "universal" analgesics, which act independently from receptor activation. Toward this end, we have designed a series of new molecules that combine the characteristics of compounds 1, 2, and 5. These new structures (Chart 2) present some of the features of molecules able to interact with G α proteins that apparently are (i) a long hydrophobic chain (which need not be peptidic); (ii) possession of at least two positive charges.³⁶

Chemistry

4,4'-Bipiperidines and 4,4'-Ethylenedipiperidines. Compounds 6–21 were synthesized as shown in Scheme 1. Commercially available 4,4'-bipiperidine dihydrochloride or 4,4'-ethylenedipiperidine dihydrochloride was

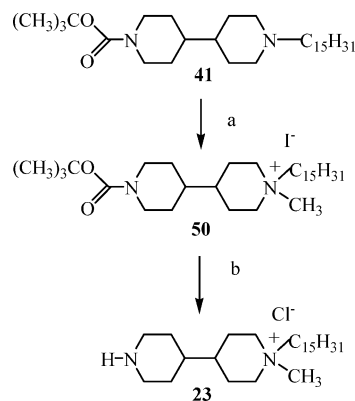
Scheme 1^a

^a Reaction conditions: (a) 10% NaOH; (b) BOC-ON; (c) bromoalkane or alkanoyl chloride, NEt_3 ; (d) 6 M HCl; (e) HCOOH, HCHO.

Scheme 2^a

^a Reaction conditions: (a) $(i\text{PrO})_4\text{Ti}$, NaBH_3CN ; (b) H_2 , 10% Pd/C.

treated with 10% NaOH to obtain the free base and then reacted with BOC-ON [2-(*tert*-butoxycarbonyloxyimino)-2-phenylacetone nitrile]²³ to give derivatives **38** and **39**. These compounds were treated with bromoalkyl derivatives and NEt_3 as a scavenger, to give **40–43**, or with the proper alkanoyl chloride, obtained from the acid with standard method,³⁷ to give **44–47**. Compounds **40–47** were then deprotected with 6 M HCl²³ to give **6–8**, **12–14**, **18**, and **20**. Finally, these compounds were treated with formic acid and formaldehyde³⁸ to give the corresponding methyl derivatives **9–11**, **15–17**, **19**, and **21**. Compound **22** has been synthesized as reported in Scheme 2. 1-Pentadecyl-piperidin-4-one **48**, obtained from 4-piperidone monohydrate hydrochloride and bromopentadecane through a synthetic pathway different from that reported in the literature,³⁹ was treated, under reductive conditions,⁴⁰ with 4-benzyl-1-amino piperidine to give **49**. Compound **49** was finally catalytically hydrogenated to obtain **22**. As shown in Scheme 3, **41**, treated with an excess of CH_3I in

Scheme 3^a

^a Reaction conditions: (a) CH_3I , Et_2O ; (b) 6 M HCl.

anhydrous ether, gave the salt **50**, which was then deprotected with 6 M HCl²³ to afford the final compound **23**. Chemical and physical characteristics of compounds **6–22** are reported in Table 1.

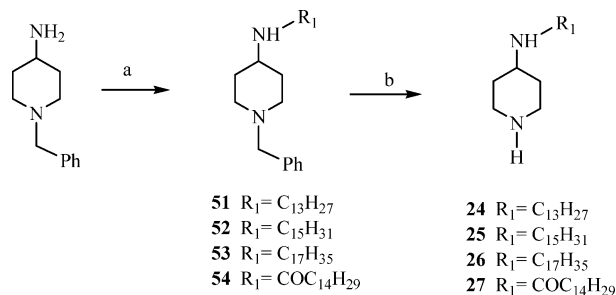
Piperidines. In Scheme 4 is shown the synthetic pathway used to obtain derivatives **24–27**. Commercially available 4-amino-1-benzylpiperidine, treated with the proper alkyl bromide or with pentadecanoyl chloride and NEt_3 , afforded alkyl derivatives **51–53** and the acyl derivative **54**, which were catalytically hydrogenated to give the final products **24–27**. As reported in Scheme 5, *N*-benzyl isonipecotic acid ethyl ester **55**,⁴¹ obtained from ethyl isonipecotate and benzyl chloride,⁴² was hydrolyzed with NaOH to give *N*-benzyl isonipecotic acid **56**. The acid was then transformed into the corresponding amide **57** using ethyl chloroformate, NEt_3 , and pentadecylamine⁴³ and hydrogenated to give **30**. Alkylation of **30** with bromoethylamine hydrobromide gave **31a** and **31b**; **31a**, after reduction with borane–methyl sulfide complex,⁴⁴ afforded compound **33**. The same reduction reaction afforded intermediate **58** from **57**. Finally, compound **32** was obtained by catalytic hydrogenation of **58**. According to Scheme 6, 1-pentadecyl-4-piperidone **48** was then transformed into **28** by reductive alkylation.⁴⁵ Compound **29** was obtained through two synthetic pathways that present similar yields despite the different number of synthetic steps. In the first attempt, 1-pentadecanoyl-4-piperidone **59**, obtained from 4-piperidone hydrochloride monohydrate with pentadecanoyl chloride, was transformed by reductive alkylation⁴⁰ into **60** that, in turn, was hydrogenated to **29** with 10% Pd/C. The same compound was more efficiently obtained from **59** by reductive alkylation with ammonium acetate.⁴⁵ Chemical and physical characteristics of compounds **24–29** and **30–33** are reported in Tables 2 and 3, respectively.

Piperazines. Compounds **34–37** were synthesized as shown in Scheme 7. Mononitrosopiperazine⁴⁶ was treated with bromopentadecane or pentadecanoyl chloride to give **61** and **62**, which were then reduced with Zn and HCl⁴⁶ to give **34** and **35**. Reduction with Zn and HCl of 4-benzyl-1-nitrosopiperazine gave 1-amino-4-benzylpiperazine **63**,⁴⁶ which was treated with bromopentadecane or pentadecanoyl chloride to give **64** and **65**, the hydrogenation of which with HCOONH_4 and 10% Pd/C in MeOH gave **36** and **37**, respectively. Chemical and physical characteristics of compounds **34–37** are reported in Table 4.

Table 1

no.	Z	R ₃	R ₁	mp (°C) base	salt	mp (°C) salt	analysis
6		H	CH ₂ -C ₁₂ H ₂₅	61–62	2HCl	>270	C ₂₃ H ₄₈ Cl ₂ N ₂
7		H	CH ₂ -C ₁₄ H ₂₉	54–55	2HCl	>290	C ₂₅ H ₅₂ Cl ₂ N ₂
8		H	CH ₂ -C ₁₆ H ₃₃	56–57	2HCl	>270	C ₂₇ H ₅₆ Cl ₂ N ₂
9		CH ₃	CH ₂ -C ₁₂ H ₂₅	<30	2HCl	>270	C ₂₄ H ₅₀ Cl ₂ N ₂
10		CH ₃	CH ₂ -C ₁₄ H ₂₉	30–31 ^a	2HCl	>270	C ₂₆ H ₅₄ Cl ₂ N ₂
11		CH ₃	CH ₂ -C ₁₆ H ₃₃	30–31 ^b	2HCl	>270	C ₂₈ H ₅₈ Cl ₂ N ₂
12		H	CO-C ₁₂ H ₂₅	56–58	HCl	150–153	C ₂₃ H ₄₅ ClN ₂ O
13		H	CO-C ₁₄ H ₂₉	43–44	HCl	167–168	C ₂₅ H ₄₉ ClN ₂ O
14		H	CO-C ₁₆ H ₃₃	43–45	HCl	165–167	C ₂₇ H ₅₃ ClN ₂ O
15		CH ₃	CO-C ₁₂ H ₂₅	<30	HCl	140–143	C ₂₄ H ₄₇ ClN ₂ O
16		CH ₃	CO-C ₁₄ H ₂₉	<30	HCl	167–169	C ₂₆ H ₅₁ ClN ₂ O
17		CH ₃	CO-C ₁₆ H ₃₃	<30	HCl	168–170	C ₂₈ H ₅₅ ClN ₂ O
18	CH ₂ CH ₂	H	CH ₂ -C ₁₄ H ₂₉	48–50	2HCl	250–253	C ₂₇ H ₅₆ Cl ₂ N ₂
19	CH ₂ CH ₂	CH ₃	CH ₂ -C ₁₄ H ₂₉	53–54	2HCl	>270	C ₂₈ H ₅₈ Cl ₂ N ₂
20	CH ₂ CH ₂	H	CO-C ₁₄ H ₂₉	30–32	HCl	162–163	C ₂₇ H ₅₃ ClN ₂ O
21	CH ₂ CH ₂	CH ₃	CO-C ₁₄ H ₂₉	42–43	HCl	95–97	C ₂₈ H ₅₅ ClN ₂ O
22	NH	H	CH ₂ -C ₁₄ H ₂₉	200–202	3HCl	>270	C ₂₅ H ₅₄ Cl ₃ N ₃

^a Compound has been purified by column chromatography using eluent B. ^b Compound has been purified by column chromatography using eluent C (see Experimental Section).

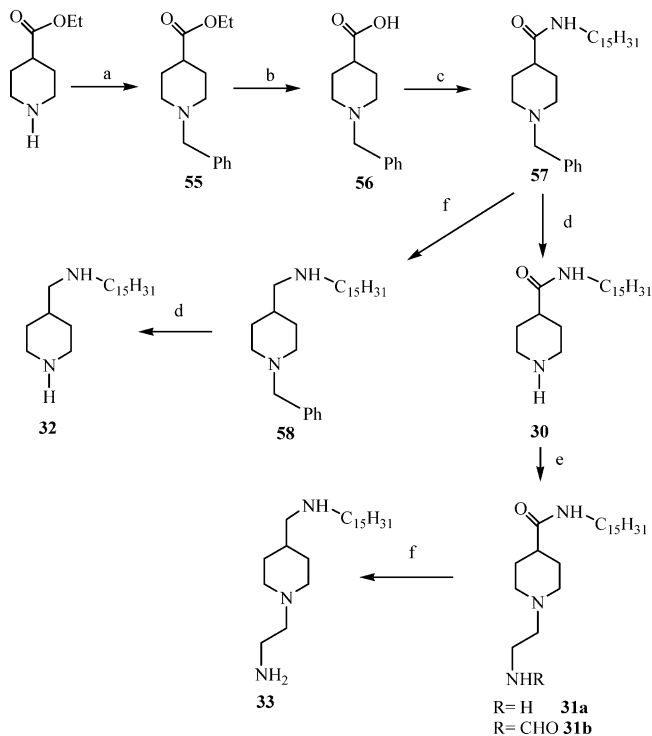
Scheme 4^a

^a Reaction conditions: (a) alkyl bromide or pentadecanoyl chloride, NEt₃; (b) H₂, 10% Pd/C.

Pharmacology

The compounds were tested for their ability to stimulate binding of GTPγS to recombinant Gi proteins, which was taken as a measure of their direct interaction with Gi proteins. Moreover, to check their Gi-activating properties we evaluated the inhibition of adenylyl cyclase activity in intact lymphocytes. Although not specific, this assay would provide information on their ability to cross the cell membrane, a critical requisite for the kind of activity we were seeking.

GTPγS(³⁵S) Binding to Recombinant Gi Proteins. Gi proteins are coupled to many receptor types and subtypes. It is also well-known that Gi proteins can be down-regulated to several levels, such as turnover and transcription, by hormones and neurotransmitters. To exclude the investigated compounds from exerting their activity in a receptor-mediated manner, the ability of the compounds to directly activate Gi proteins was evaluated on recombinant Gi isoforms. Recombinant Gα_{i/o} subunits were reconstituted in an *in vitro* system, and GTPγS binding was measured to verify the protein activation level in the presence of the tested compounds. To unmask a potential selectivity of these molecules toward Gi protein isoforms, α_{i1}, α_{i3}, and α_{o1} subunits were expressed and purified. Binding of GTPγS (the nonhydrolyzable GTP analogue) was then measured in the presence of our compounds that, if endowed with Gi protein stimulatory activity, would increase the basal

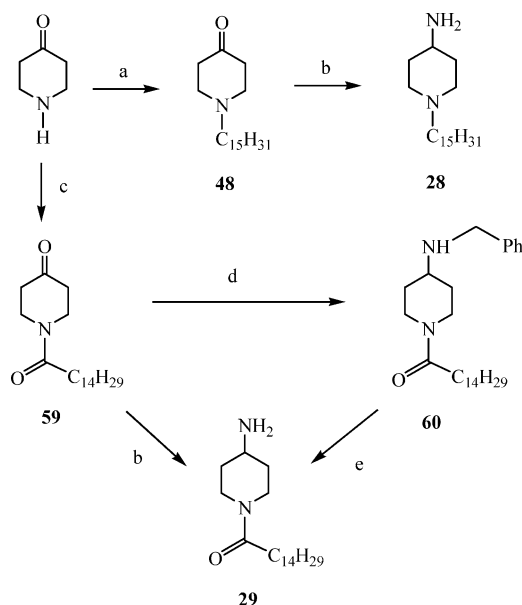
Scheme 5^a

^a Reaction conditions: (a) benzyl chloride, NEt₃; (b) NaOH; (c) EtOCOCl, NEt₃, pentadecylamine; (d) H₂, 10% Pd/C; (e) bromoethylamine; (f) (CH₃)₂S·BH₃.

binding of GTPγS. **1** and **5** were used as reference compounds.

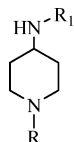
Gi Protein Modulation. Inhibition of adenylyl cyclase activity in intact lymphocytes from human peripheral blood was measured, taking into account that Gi proteins have a ubiquitous distribution.⁴⁷ Lymphocytes represent an appropriate tool to detect Gi protein functionality because they have cell surface recognition sites for pertussis toxin, the selective inactivator of Gi. These recognition sites are lacking in other blood elements such as erythrocytes.⁴⁸

Enzymatic activity was previously amplified by forskolin, a diterpene that selectively activates adenylyl

Scheme 6^a

^a Reaction conditions: (a) bromopentadecane, NEt₃; (b) CH₃-COONH₄, NaBH₃CN; (c) pentadecanoyl chloride, NEt₃; (d) benzylamine, (iPrO)₄Ti, NaBH₃CN; (e) H₂, 10% Pd/C.

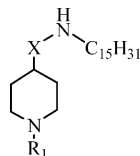
Table 2



no.	R	R ₁	mp (°C)		analysis
			base	salt	
24	H	CH ₂ -C ₁₂ H ₂₅	30–31 ^a	2HCl	268–269 C ₁₈ H ₄₀ Cl ₂ N ₂
25	H	CH ₂ -C ₁₄ H ₂₉	42–43	2HCl	245–246 C ₂₀ H ₄₄ Cl ₂ N ₂
26	H	CH ₂ -C ₁₆ H ₃₃	30–32 ^b	2HCl	250–251 C ₂₂ H ₄₈ Cl ₂ N ₂
27	H	CO-C ₁₄ H ₂₉		HCl	106–108 C ₂₀ H ₄₁ ClN ₂ O
28	CH ₂ -C ₁₄ H ₂₉	H	<i>a</i>	2HCl	>240 C ₂₀ H ₄₄ Cl ₂ N ₂
29	CO-C ₁₄ H ₂₉	H	50–51 ^a	HCl	210–212 C ₂₀ H ₄₁ ClN ₂ O

^a Compound has been purified by column chromatography using eluent B. ^b Compound has been purified by column chromatography using eluent D (see Experimental Section).

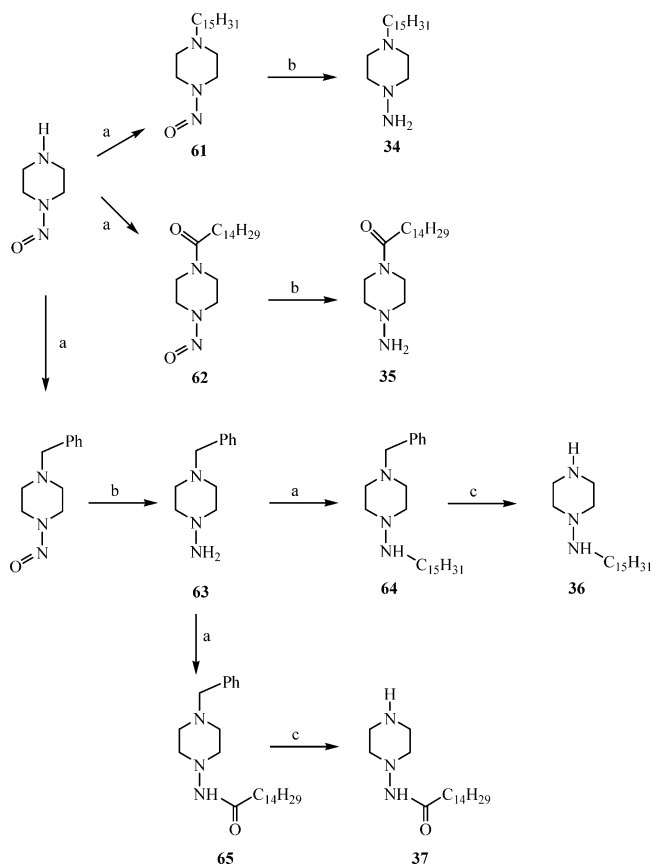
Table 3



no.	R ₁	X	mp (°C)		analysis
			base	salt	
30	H	CO	73–74	HCl	220–221 C ₂₁ H ₄₃ ClN ₂ O
31a	CH ₂ CH ₂ NH ₂	CO	100–101	2HCl	218–220 C ₂₃ H ₄₉ Cl ₂ N ₃ O
31b	CH ₂ CH ₂ NHCHO	CO	109–110	HCl	201–203 C ₂₄ H ₄₈ ClN ₃ O ₂
32	H	CH ₂	31–33	2HCl	222–225 C ₂₁ H ₄₆ Cl ₂ N ₂
33	CH ₂ CH ₂ NH ₂	CH ₂	40–43 ^a	3HCl	215–220 C ₂₃ H ₅₂ Cl ₃ N ₃

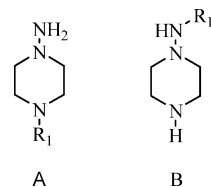
^a Compound has been purified by column chromatography using eluent D (see Experimental Section).

cyclase,⁴⁹ at a concentration of 10⁻⁴ M, which is reported to be the most effective one in human lymphocytes.⁵⁰ The forskolin-stimulated adenylyl cyclase activity was inhibited, with similar potency, by the nonhydrolyzable GTP analogue Gpp(NH)p, used as physiological control,

Scheme 7^a

^a Reaction conditions: (a) alkyl bromide or alkanoyl chloride, NEt₃; (b) Zn, HCl; (c) HCOONH₄, 10% Pd/C, MeOH.

Table 4



no.	structure	R ₁	mp (°C)		analysis
			base	salt	
34	A	CH ₂ -C ₁₄ H ₂₉	30–32 ^a	3HCl	217–218 C ₁₉ H ₄₄ Cl ₃ N ₃
35	A	CO-C ₁₄ H ₂₉	35–36 ^a	2HCl	195–197 C ₁₉ H ₄₁ Cl ₂ N ₃ O
36	B	CH ₂ -C ₁₄ H ₂₉	28–30	3HCl	240–241 C ₁₉ H ₄₄ Cl ₃ N ₃
37	B	CO-C ₁₄ H ₂₉	<30	2HCl	122–123 C ₁₉ H ₄₁ Cl ₂ N ₃ O

^a Compound has been purified by column chromatography using eluent B (see Experimental Section).

and by the known Gi-protein activators **1**²¹ and **5**,²³ used as reference compounds. Considering the multiplicity of GTP-binding proteins present in lymphocytes,⁵¹ it may seem difficult to evidence a particular G-protein subtype activation with the use of a GTP analogue. However, not only a preferential affinity of Gpp(NH)p toward Gi proteins in the range of concentrations used has been evidenced,^{52,53} but also a widely different expression of each type of G protein is documented, indicating that Gi are commonly 10 times more abundant than Gs.⁵⁴ Therefore, Gpp(NH)p appears to be a suitable tool to reproduce a physiological Gi protein activation. Preincubation of lymphocytes with pertussis toxin at a concentration consistent with a complete ADP-ribosylation of Gi proteins⁵⁵ prevented the inhibi-

tion of the forskolin-stimulated adenylyl cyclase activity (data not shown), suggesting that the effects observed were related to a Gi protein system activation.

Results

GTP γ S(³⁵S) Binding to Recombinant Gi Proteins. Gi protein activation induces the exchange of GDP for GTP on the α subunit. As a result, α -GTP dissociates from $\beta\gamma$ subunits. Therefore, the evaluation of GTP γ S binding to the α subunit is a good method to ascertain the ability of tested compounds to activate Gi protein. Stimulation of binding at 5×10^{-5} M (percent of basal/30 min) was taken as a measure of efficacy and EC₅₀ (determined using concentrations of 1×10^{-4} , 5×10^{-5} , 3×10^{-5} , 2×10^{-5} , 1×10^{-5} , 5×10^{-6} , 1×10^{-6} , and 1×10^{-7} M of each compound) as a measure of potency (Table 5). To compare the efficacy of the tested molecules we chose the maximum effect obtainable at the concentration of 5×10^{-5} M, because higher concentrations, although more active, showed a higher aspecific binding. To test the method specificity we first evaluated the effect of (i) a compound (morphine) that activates Gi protein in a receptor-mediated manner, (ii) a compound (atenolol) that blocks receptor-mediated Gs activation, and (iii) a compound (AMPA) that modulates Na⁺ channels. All three compounds were completely inactive (data not shown). As far as reference compound **1** is concerned, we confirmed its effect on α_{i1} and α_{o1} subunits and demonstrated its activity also on α_{i3} . Instead, reference compound **5**, which had never been tested before by this method, showed little activity on α_{i1} and α_{o1} but was able to stimulate GTP γ S binding on α_{i3} (Table 5). Figure 1 shows a comparison of the effects of increasing concentrations of compound **1** and compounds **34** and **36** on G α_{i1} (panel A), G α_{i3} (panel B), and G α_{o1} (panel C) isoforms. Among the newly synthesized compounds, comparing the effect at 5×10^{-5} M, we found that compounds **8**, **18**, **22**, **25**, **28**, **31a**, **31b**, **32**, **34**, **36**, and **37** were superior to compound **1** in increasing GTP γ S binding. Interestingly, compounds **7**, **18**, **30**, and **32** were particularly active on G α_o , and compound **8** was notably active on G α_i . Compounds **7**, **13**, **23**, **24**, **27**, **29**, and **30** showed an activity roughly comparable to that of **1**, whereas compounds **6**, **9**, **12**, **14**, **16**, **17**, **20**, **21**, and **35** were less active than **1**. All active compounds exerted their effect at EC₅₀ > 1×10^{-6} M (Table 5). Compounds **10**, **11**, **15**, and **19** were completely inactive.

Protein Gi Activation. Adenylyl cyclase activity was stimulated by using the diterpene forskolin at a concentration of 10^{-4} M. A relative value of 100% (100.1 ± 4.7) was assigned to the concentration of cAMP obtained after forskolin stimulation. The inhibition of the stimulated cAMP production by Gi proteins was then evaluated by administration of Gpp(NH)p, a nonhydrolyzable GTP analogue used as positive control, in the concentration range of 10^{-10} – 10^{-4} M. The results, reported in Figure 2, are expressed as percent of stimulated cAMP production relative to control (10^{-4} M forskolin). Gpp(NH)p produced a dose-dependent inhibition of forskolin-stimulated adenylyl cyclase activity detectable by a reduction of cAMP levels of ~35% (Figure 2A). Similarly to Gpp(NH)p, also the reference compounds **1** and **5** produced a dose-dependent reduction of cAMP

levels (Figure 2A). Similar results were obtained when the experiments were carried out on permeabilized lymphocytes (data not shown). At the concentration of 10^{-10} M they were both inactive; statistical significance was reached at 10^{-7} M, and their maximum effect was at 10^{-5} M. The effect observed was prevented by pretreatment with pertussis toxin, indicating that the cAMP level reduction is due to a Gi protein system activation (data not shown). Among the molecules investigated, different pharmacological profiles were observed (Figure 2). Compounds **7**, **9**, and **36** inhibited cAMP production, reaching statistical significance at 10^{-7} M (Figure 2B,C). The intensity of the effect produced was similar to that produced by the reference drugs **1** and **5**. Compound **7** appeared to be one of the most active, but its chloromethyl derivative, **23**, which carries a permanent charge on the nitrogen, showed a much lower ability to reduce the cAMP levels (Figure 2C). Compounds **24** and **34** were able to inhibit the cAMP production with an efficacy similar to that of the reference drugs, but the statistical significance was reached at 10^{-6} M (Figure 2D). A reduction of the cAMP production was also observed with compounds **13**, **29**, and **31b**, which showed statistical significance at 10^{-5} M (Figure 2E). Finally, a Gi protein-activating effect was also observed for compounds **16**, **18**, **20**–**22**, **25**, **30**, and **32** even if the reduction of cAMP production was significant only at 10^{-4} M (Figure 2F).

Compounds **6**, **8**, **10**–**12**, **14**, **15**, **17**, **19**, **27**, **28**, **31a**, **33**, **35**, and **37** were unable to reduce the cAMP production at all concentrations tested (data not shown). Among them, compounds **27**, **28**, and **31a** were unable to produce any modification on cAMP levels also in permeabilized cells, ruling out the possibility that their inactivity was due to a poor ability to cross the plasma membranes (data not shown).

Discussion

Compounds **7**, **9**, **13**, **18**, **20**–**22**, **24**, **25**, **29**, **30**, **31b**, **32**, **34**, and **36** were able to stimulate GTP γ S binding and to inhibit cAMP production. The efficacy of the above-mentioned compounds in inhibiting the adenylyl cyclase activity was similar to that of the reference drugs because the highest reduction of cAMP production was reached at 10^{-5} – 10^{-4} M, in correspondence to the maximum effect produced by Gpp(NH)p. Among them, compounds **7**, **9**, **24**, **34**, and **36** showed the highest potency, being active at 10^{-7} – 10^{-6} M. Therefore, in agreement with the good correlation between the results of the two methods, it is very likely that these compounds act through a direct activation of Gi proteins. This is also suggested by the behavior of **23**, the chloromethylated form of **7** (see Scheme 3) that, having a permanent charge on the nitrogen, should cross less efficiently cell membranes, at least up to 10^{-5} M. As a matter of fact, **23** shows a much lower activity on lymphocytes than **7**, which is able to reduce cAMP production at concentrations 1000 times lower. In support of such a hypothesis, **23** showed an activity similar to that of **7** on GTP γ S binding where, the test being performed on isolated Gi/o proteins, no membrane crossing is necessary. In this respect, it can be hypothesized that a specific polyamine transporter might be involved in the process. At the moment, we have no

Table 5. Stimulation of [³⁵S]GTPγS Binding to Gi Protein Isoforms by Compounds **6–37**

[³⁵ S]GTPγS binding on Gα _{i/o} subunit							
Gα isoform				basal/30 min			
	α _{i1}						
	α _{i3}						
	α _{o1}						
[³⁵ S]GTPγS binding on Gα _{i/o} subunit							
no.	Gα isoform	stimulation at 5 × 10 ⁻⁵ M		no.	Gα isoform	stimulation at 5 × 10 ⁻⁵ M	
		% of basal/30 min ^a	EC ₅₀ (μM)			% of basal/30 min ^a	EC ₅₀ (μM)
1	α _{i1}	900 ± 85*	47.9 ± 2.0	22	α _{i1}	1539 ± 57*	32.4 ± 14.7
	α _{i3}	1030 ± 137*	43.9 ± 1.4		α _{i3}	2275 ± 131*	14.4 ± 0.3
	α _{o1}	2510 ± 118*	28.8 ± 0.7		α _{o1}	3037 ± 172*	13.5 ± 1.7
5	α _{i1}	60 ± 7	38.8 ± 6.4	23	α _{i1}	1576 ± 187*	14.7 ± 0.7
	α _{i3}	391 ± 34*			α _{i3}	1626 ± 150*	16.0 ± 0.3
	α _{o1}	129 ± 22			α _{o1}	2086 ± 205*	10.1 ± 0.3
6	α _{i1}	755 ± 45*	21.1 ± 1.7	24	α _{i1}	1484 ± 166*	36.4 ± 13.5
	α _{i3}	406 ± 29*	21.2 ± 4.0		α _{i3}	2005 ± 97*	18.8 ± 2.9
	α _{o1}	913 ± 96*	27.1 ± 16.1		α _{o1}	2349 ± 204*	25.5 ± 5.2
7	α _{i1}	1510 ± 166*	19.1 ± 7.6	25	α _{i1}	3368 ± 173*	11.8 ± 1.3
	α _{i3}	1640 ± 82*	18.7 ± 6.8		α _{i3}	3443 ± 263*	11.2 ± 0.9
	α _{o1}	1810 ± 133*	6.2 ± 1.1		α _{o1}	3826 ± 218*	13.3 ± 1.4
8	α _{i1}	5694 ± 173*	35.7 ± 7.1	26^b	α _{i1}		
	α _{i3}	5658 ± 225*	41.5 ± 10.9		α _{i3}	nd ^c	nd
	α _{o1}	3476 ± 345*	13.4 ± 0.1		α _{o1}		
9	α _{i1}	704 ± 55*	18.7 ± 2.5	27	α _{i1}	1270 ± 76*	50.2 ± 11.0
	α _{i3}	435 ± 26*	18.9 ± 6.6		α _{i3}	1287 ± 64*	34.3 ± 7.1
	α _{o1}	420 ± 11*	19.3 ± 2.4		α _{o1}	1366 ± 216*	22.7 ± 4.4
10	α _{i1}	97 ± 14	-	28	α _{i1}	3346 ± 97*	22.1 ± 2.7
	α _{i3}	80 ± 8			α _{i3}	3775 ± 69*	27.6 ± 2.5
	α _{o1}	93 ± 21			α _{o1}	6893 ± 79*	23.0 ± 1.4
11	α _{i1}	81 ± 10		29	α _{i1}	1051 ± 69*	25.5 ± 6.7
	α _{i3}	97 ± 14			α _{i3}	1071 ± 56*	25.1 ± 9.3
	α _{o1}	99 ± 22			α _{o1}	1279 ± 53*	25.8 ± 6.9
12	α _{i1}	950 ± 75*	17.6 ± 3.9	30^c	α _{i1}	973 ± 314*	6.4 ± 0.2
	α _{i3}	1178 ± 24*	12.1 ± 2.8		α _{i3}	986 ± 175*	5.9 ± 0.5
	α _{o1}	1498 ± 68*	18.1 ± 0.8		α _{o1}	2271 ± 254*	8.2 ± 1.3
13	α _{i1}	1542 ± 225*	8.6 ± 1.0	31a	α _{i1}	2629 ± 159*	18.4 ± 3.7
	α _{i3}	1558 ± 125*	9.3 ± 1.4		α _{i3}	2724 ± 103*	19.0 ± 3.7
	α _{o1}	2388 ± 215*	5.6 ± 0.5		α _{o1}	3347 ± 322*	16.9 ± 2.6
14	α _{i1}	612 ± 104*	25.4 ± 2.0	31b	α _{i1}	3291 ± 128*	8.6 ± 1.0
	α _{i3}	424 ± 131*	24.4 ± 1.4		α _{i3}	3244 ± 275*	9.3 ± 1.4
	α _{o1}	920 ± 248*	22.6 ± 0.4		α _{o1}	3597 ± 335*	5.5 ± 0.4
15	α _{i1}	84 ± 7		32	α _{i1}	2464 ± 179*	10.6 ± 0.4
	α _{i3}	75 ± 15			α _{i3}	1933 ± 133*	11.4 ± 0.4
	α _{o1}	126 ± 23			α _{o1}	4162 ± 481*	12.4 ± 2.8
16	α _{i1}	392 ± 32*	6.0 ± 0.5	33	α _{i1}	825 ± 105*	17.8 ± 2.5
	α _{i3}	265 ± 31*	17.9 ± 2.8		α _{i3}	815 ± 65*	23.0 ± 6.8
	α _{o1}	531 ± 26*	34.9 ± 4.1		α _{o1}	942 ± 143*	24.3 ± 7.3
17	α _{i1}	232 ± 48*	23.0 ± 2.9	34^d	α _{i1}	2328 ± 334*	6.5 ± 0.9
	α _{i3}	181 ± 72	23.3 ± 4.1		α _{i3}	2816 ± 351*	6.5 ± 0.3
	α _{o1}	472 ± 63*	21.4 ± 0.8		α _{o1}	2870 ± 565*	3.7 ± 0.3
18	α _{i1}	2457 ± 210*	451 ± 988	35	α _{i1}	173 ± 14*	28.6 ± 0.5
	α _{i3}	2404 ± 262*	8050.3 ± 9260		α _{i3}	211 ± 51*	24.6 ± 3.0
	α _{o1}	5009 ± 415*	10.5 ± 2.7		α _{o1}	194 ± 47*	373.9 ± 292
19	α _{i1}	86 ± 13		36^d	α _{i1}	5060 ± 463*	15.9 ± 3.2
	α _{i3}	102 ± 8	-		α _{i3}	5718 ± 293* ^f	19.0 ± 2.9
	α _{o1}	132 ± 16	-		α _{o1}	5577 ± 479*	6.1 ± 0.5
20	α _{i1}	845 ± 141*	6.6 ± 0.1	37	α _{i1}	2322 ± 221*	38.5 ± 3.4
	α _{i3}	1093 ± 131*	6.3 ± 0.3		α _{i3}	3690 ± 280*	24.9 ± 1.1
	α _{o1}	724 ± 293*	6.1 ± 0.1		α _{o1}	4636 ± 292*	23.4 ± 2.2
21	α _{i1}	577 ± 40*	20.1 ± 6.0				
	α _{i3}	815 ± 66*	6.1 ± 0.4				
	α _{o1}	742 ± 43*	12.5 ± 0.1				

^a These values are taken as a measure of efficacy in stimulating binding. *, *P* < 0.01. ^b Compound does not solubilize. ^c Compound is sparingly soluble at 10⁻⁵ M. ^d Compound is sparingly soluble at 10⁻⁴ M. ^e Not done. ^f Experiment is performed at 3 × 10⁻⁵ M.

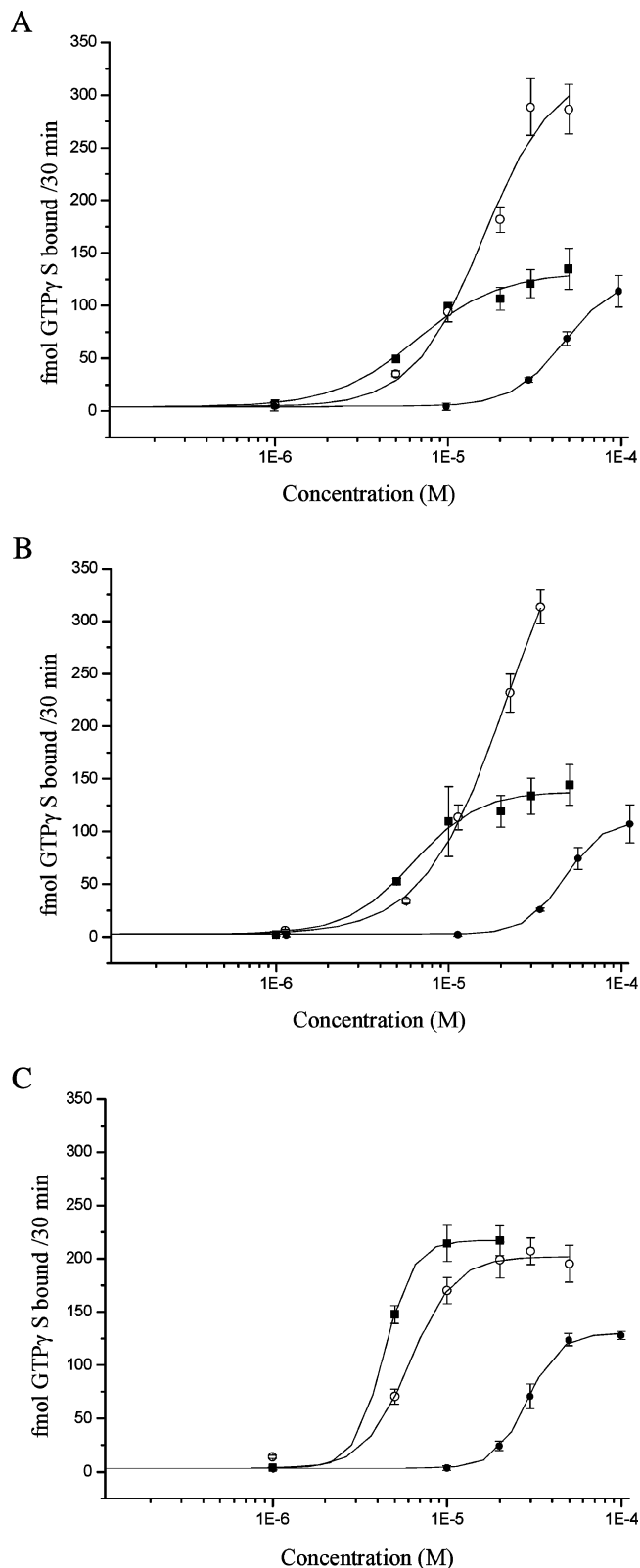


Figure 1. Effect of increasing concentrations of compounds **1** (●), **34** (■), and **36** (○) on the stimulation of [³⁵S]GTP_γS binding on G_{α_{i1}} (A), G_{α_{i3}} (B), and G_{α_{o1}} (C). Data are expressed as mean ± SEM.

indication that this is indeed the case, but this possibility cannot be ruled out.

Compounds **6**, **8**, **12**, **14**, **17**, **27**, **28**, **31a**, **33**, and **37** were inactive in inhibiting cAMP production in lymphocytes but effective in stimulating GTP_γS binding. This discrepancy might be due to the lack of selectivity

of these compounds that, besides Gi proteins, could interact with other targets that would oppose the Gi protein-inhibiting effect. For instance, they might interact with other G protein subtypes, such as Gs proteins, producing a stimulation of adenylyl cyclase which, in its turn, would lead to an increase of the intracellular cAMP contents. It should also be taken into account that their lack of activity on cells might be due to their difficulties in crossing the cell membrane or their poor solubility. On the basis of the results obtained in both assays, the most interesting compounds of the series appear to be **7**, **9**, **22**, **25**, **30**, **31b**, **32**, **34**, and **36**, which seem to be able to cross the cell membrane and show good efficacy and potency in the low micromolar range. Among them, compound **34** appears to be the most potent and **36** the most efficacious. Some subtype selectivity also seems to be present in some of these compounds, in particular **7**, **18**, **30**, and **32**, which show almost a 2-fold efficacy on the α_o subunit with respect to the others.

Establishing structure–activity relationships (SAR) was not the main goal of this research, yet a few correlations between the structure of our compounds and Gi-activating properties can be made. First of all, we were able to confirm some of the molecular requisites that have already been proposed for drugs activating Gi proteins.^{21,36} As a matter of fact, with the exceptions of **30** and **31b**, the most active compounds present two or three basic amino groups that are very likely protonated under physiological conditions. In this respect, it is interesting to note that although **31a** (two basic centers) and **31b** (one basic center) are equally active in facilitating GTP_γS binding, **31a** does not present any activity on cells, a finding that would suggest that it has problems in crossing the membrane. However, cell permeabilization did not produce any modification on cAMP levels, thus ruling out this possibility. At the moment, we do not have a sound explanation for the different behaviors of **31a** and **31b**. It is also important to note that a secondary terminal nitrogen seems to be preferable for Gi-modulating activity as *N*-methyl derivatives present little or no activity and none of them is present in the list of the most interesting compounds. A second feature, shared by our compounds with drugs known to act directly on Gi proteins, is the presence of a long lipophilic carbon chain. In this respect, even if the length of the methylene chains used in the present research did not vary much, the *n*-pentadecyl chain seems to be the most useful to induce Gi-activating properties.

We found active compounds in all four groups of structures synthesized, which vary essentially in the relative position of the basic nitrogen atoms. This seems to suggest that the arrangement of the basic nitrogen atoms is not crucial for activity, whereas it may contribute to subtype selectivity. However, other research and more dedicated SAR studies will be necessary to confirm these aspects.

Going briefly into details, among the 4,4'-piperidinylo piperidine derivatives, compound **7** is the most active, those with an amido group or with shorter methylene chains being inactive or feebly active. An interesting feature of **7** is its selectivity toward the α_o subtype, in both efficacy and potency. On the contrary, compound

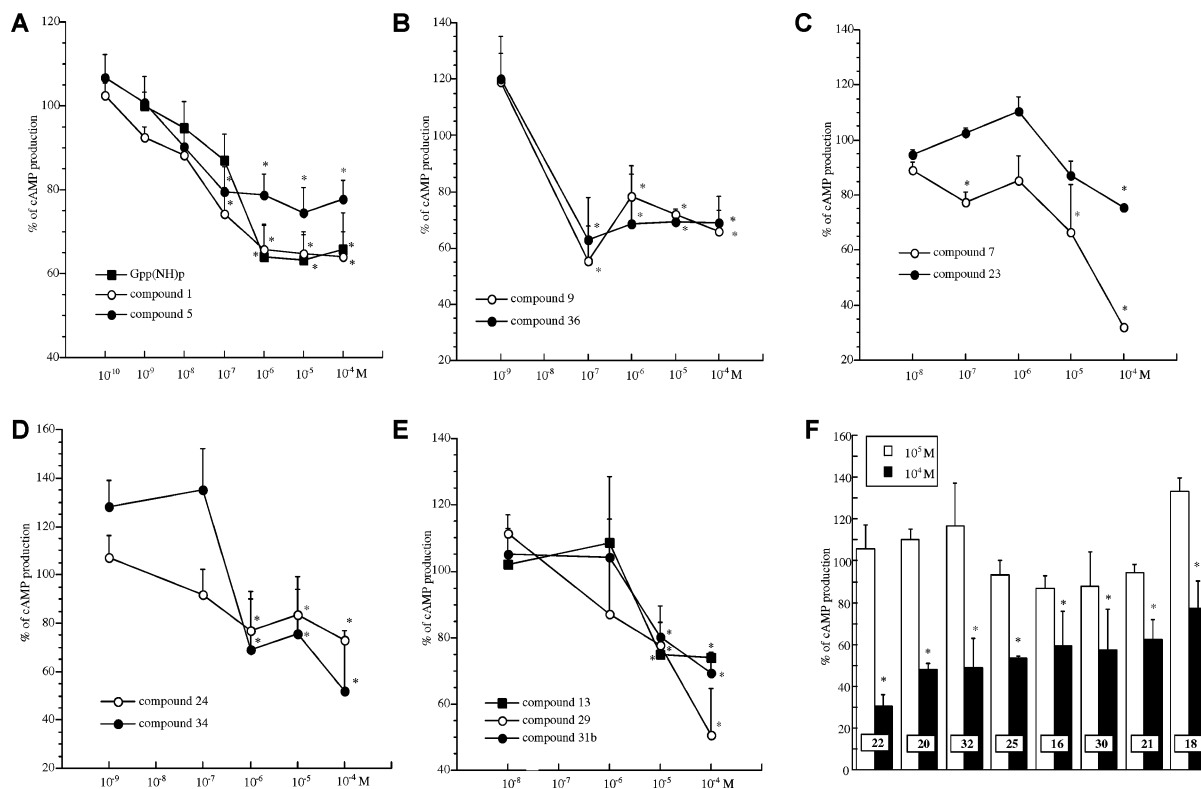


Figure 2. Evaluation of the capability to inhibit the stimulated adenylyl cyclase activity by the investigated compounds. Adenylyl cyclase activity was stimulated by 10^{-4} M forskolin: (A) dose-dependent reduction of cAMP production by the reference drugs [Gpp(NH)p, compounds **1** and **5**]; (B) effect on cAMP production by compounds **9** and **36**; (C) effect on cAMP production by compounds **7** and **23**; (D) effect on cAMP production by compounds **24** and **34**; (E) effect on cAMP production by compounds **13**, **29**, and **31b**; (F) effect on cAMP production by compounds **22**, **20**, **32**, **25**, **16**, **30**, **21**, and **18**. Data are expressed as mean \pm SEM relative to control (10^{-4} M forskolin), which was assigned a relative value of 100% (100.1 ± 4.7). Each data point represents the mean of at least three independent experiments. Experiments were performed in duplicate. *, $P < 0.05$ in comparison with forskolin-stimulated lymphocytes.

8, which carries a longer methylene chain, is one of the most efficacious in activating $GTP\gamma S$ binding but shows the opposite selectivity and does not activate G_i proteins in lymphocytes. Introduction of a NH group between the piperidine rings (**22**) maintained activity, whereas introduction of a two-methylene chain (**18**) reduced potency but maintained selectivity for the α_o subtype.

Among the 4-aminopiperidine derivatives (Table 2) compound **25** is the most interesting, in terms of both potency and efficacy; even in this case, reduction (**24**) of the length of the methylene chain reduces activity. The extended chain derivative **26** could not be tested because of solubility problems.

Among the derivatives of piperidine-4-carboxylic acid (Table 3) is found the only active compound possessing only one basic nitrogen (**30**), which combines activity and selectivity toward the α_o subtype. However, confirming that basic nitrogen atoms are important for activity, reduction of the carbonyl group of **30** to give **32** increases activity with little impact on selectivity. Accordingly, introduction of an ethylamino group as in **31a** and **31b** results in two compounds active in stimulating $GTP\gamma S$ binding.

Finally, in the *N*-aminopiperazine group (Table 4) are found two of the most interesting drugs of the series. Compound **34** shows a good efficacy and is the most potent compound studied, particularly on the α_o subtype; its regioisomer, **36**, is the most efficacious. Both, however, do not show selectivity, suggesting that this

arrangement of the basic nitrogen atoms is less useful than the other ones to exalt this property.

In conclusion, we have identified a number of compounds with different chemical structures that seem to be able to directly activate G_i proteins and that will be useful for developing new pharmacological tools to study these essential players in the transmission of information between cells and, eventually, to develop clinically useful drugs.

Experimental Section

Chemistry. All melting points were taken on a Büchi apparatus and are uncorrected. Infrared spectra were recorded with a Perkin-Elmer 681 spectrophotometer in Nujol mull for solids and neat for liquids. Unless otherwise stated, NMR spectra were recorded on a Gemini 200 spectrometer (200 MHz for 1H NMR, 50.3 MHz for ^{13}C), and chromatographic separations were performed on a silica gel column by gravity chromatography (Kieselgel 40, 0.063–0.200 mm; Merck) or flash chromatography (Kieselgel 40, 0.040–0.063 mm; Merck). When necessary, chromatographic separations were performed on an Al_2O_3 column by gravity chromatography (aluminum oxide 90 standardized, Merck). Yields are given after purification, unless otherwise stated. Where analyses are indicated by symbols, the analytical results are within $\pm 0.4\%$ of the theoretical values. Eluents used in the purification step were (A) petroleum ether/ Et_2O / CH_2Cl_2 /absolute $EtOH/NH_4OH$, 450:180:180:45:2.5; (B) CH_2Cl_2 /absolute $EtOH$ /petroleum ether/ NH_4OH , 340:65:60:8; (C) petroleum ether/ Et_2O / CH_2Cl_2 /absolute $EtOH/NH_4OH$, 900:360:360:180:9.9; and (D) CH_2Cl_2 / $MeOH/NH_4OH$, 50:50:3. Melting points, yields, and column eluents of intermediates are reported in Table 6.

Table 6

no.	mp (°C) base ^a	yield (%)	column eluent
38 ^b		45	
39		42	D
40	30–31	62	A
41	33–32	61	A
42	30–31	64	A
43	42–43	41	A
44	35–37	70	A
45	31–33	63	A
46	35–38	86	A
47	33–34	55	A
48 ^c	30–31	49	C
49	34–35	64	B
50	140–142 ^d	99	
51	oil	99	A and C
52	<30	62	B
53	<30	76	A and C
54	oil	94	
55	oil	92	
56 ^e	oil	91	
57	47–50	56	A
58	50–51	71	A and B
59	oil	74	C
60	oil	69	C
61	31–32	21	B
62	56–57	85	C
63 ^f	oil	33	B
64	30–32	70	A
65	<30	84	B

^a Intermediates were identified through IR and NMR spectra; microanalyses were not performed. ^b See ref 23. ^c See ref 39. ^d Melting point of the iodomethylate. ^e See ref 41. ^f See ref 46.

tert-Butyl-(1'-tridecyl)-[4,4']-bipiperidinyl-1-carboxylate (40). To *tert*-butyl-[4,4']-bipiperidinyl-1-carboxylate **38**²³ in anhydrous DMF (28 mL × g) were added 2 equiv of NEt₃ and 2 equiv of bromotridecane. After 8 h at reflux, the solvent was removed, the residue made alkaline with NaHCO₃, and the solution extracted with CH₂Cl₂. Following anhydrication and evaporation of the solvent, the residue was purified by column chromatography using eluent A to obtain **40** in 62% yield: IR (Nujol) ν 1691 (OCN) cm⁻¹; ¹H NMR (CDCl₃) δ 0.88 (t, *J* = 6.6 Hz, 3H, CH₃CH₂), 1.06–1.21 (m, 6H, CH₂ bipiperidine), 1.21–1.37 (m, 20H, 10CH₂C), 1.35–1.56 (m, 2H, CH₂CH₂N), 1.46 (s, 9H, CH₃C), 1.60–1.75 (m, 4H, CH₂ bipiperidine + 2CH), 1.75–1.91 (m, 2H, CH₂N bipiperidine), 2.28 (t, *J* = 7.7 Hz, 2H, CH₂CH₂N), 2.55–2.72 (m, 2H, 2CH_{eq}NBOC), 2.91–3.04 (m, 2H, CH₂N bipiperidine), 4.04–4.19 (m, 2H, 2CH_{ax}NBOC). Anal. (C₂₈H₅₄N₂O₂) C, H, N.

With the same method described for **40**, starting from **38** or **39** and the proper bromoalkane or alkanoyl chloride, compounds **41–47** were obtained.

1-Tridecyl-[4,4']-bipiperidine (6). Two equivalents of 6 N HCl was added to **40** dissolved in a few milliliters of ethyl acetate and left for 12 h at room temperature. The mixture was basified with 10% NaOH in H₂O and extracted with CH₂Cl₂; anhydrication and evaporation of the solvent gave **6** in quantitative yield: IR (Nujol) ν 3230 (NH) cm⁻¹; ¹H NMR (CDCl₃) δ 0.88 (t, *J* = 6.9 Hz, 3H, CH₃CH₂), 1.01–1.21 (m, 6H, CH₂ bipiperidine), 1.21–1.40 (m, 20H, 10CH₂C), 1.41–1.56 (m, 2H, CH₂CH₂N), 1.61–1.75 (m, 5H, CH₂ bipiperidine + 2CH + NH), 1.75–1.90 (m, 2H, CH₂N bipiperidine), 2.27 (t, *J* = 7.9 Hz, 2H, CH₂CH₂N), 2.49–2.64 (m, 2H, CH₂NH), 2.90–3.03 (m, 2H, CH₂N bipiperidine), 3.03–3.15 (m, 2H, CH₂NH); ¹³C NMR (CDCl₃) δ 14.53 (q, CH₃), 23.10 (t, CH₂), 27.53 (t, CH₂), 28.20 (t, CH₂), 29.71 (t, CH₂), 30.02 (t, CH₂), 31.04 (t, CH₂), 32.32 (t, CH₂), 41.75 (d, CH), 41.84 (d, CH), 47.45 (t, CH₂), 54.77 (t, CH₂), 59.74 (t, CH₂). Anal. (C₂₃H₄₆N₂) C, H, N.

Compounds **7**, **8**, **12–14**, **18**, and **20** were obtained in quantitative yields using the same method as described for **6**.

Compounds **6–8**, **12–14**, **18**, and **20** were transformed in the hydrochlorides by treatment with EtOH/HCl in Et₂O anhydrous. The salts were recrystallized from absolute EtOH and anhydrous Et₂O.

1-Methyl-1'-tridecyl-[4,4']-bipiperidine (9). Formic acid (17 equiv) and 40% formaldehyde in H₂O (4.8 equiv) were added to compound **6** in EtOH and refluxed for 4 h. The solution was made alkaline with 10% NaOH and extracted with CH₂Cl₂. After anhydrication, evaporation of the solvent gave **9** as a pure compound: ¹H NMR (CDCl₃) δ 0.89 (t, *J* = 6.9 Hz, 3H, CH₃CH₂), 0.96–1.15 (m, 2H, CH₂ bipiperidine), 1.15–1.41 (m, 24H, 10CH₂C + 2CH₂ bipiperidine), 1.41–1.56 (m, 2H, CH₂CH₂N), 1.61–1.76 (m, 4H, CH₂ bipiperidine + 2CH), 1.76–1.92 (m, 4H, 2CH₂N bipiperidine), 2.21–2.32 (m, 2H, CH₂CH₂N), 2.25 (s, 3H, CH₃N), 2.80–2.90 (m, 2H, CH₂N bipiperidine), 2.90–3.01 (m, 2H, CH₂NCH₃); ¹³C NMR (CDCl₃) δ 14.54 (q, CH₃), 23.10 (t, CH₂), 27.54 (t, CH₂), 28.20 (t, CH₂), 29.77 (t, CH₂), 29.91 (t, CH₂), 29.97 (t, CH₂), 30.04 (t, CH₂), 32.33 (t, CH₂), 40.66 (d, CH), 41.33 (d, CH), 46.88 (q, CH₃), 54.75 (t, CH₂), 56.68 (t, CH₂), 59.74 (t, CH₂). Anal. (C₂₄H₄₇N₂) C, H, N.

Using the same method as described above and starting from **7**, **8**, **12–14**, **18**, and **20**, derivatives **10**, **11**, and **15–17** were obtained, respectively. Compounds **10** and **11** were purified by column chromatography using eluents B and C, respectively, whereas compounds **15–17**, **19**, and **21** were obtained pure.

Compounds **9–11**, **15–17**, **19**, and **21** were transformed into the hydrochlorides with the method described above.

(1-Benzyl-4'-piperinyl)-(1-pentadecyl-4'-piperidinyl)-amine (49). One equivalent of **48**, 1 equiv of 4-amino-1-benzylpiperidine, and 1.1 equiv of (iPrO)₄Ti were left at room temperature until the IR carbonyl absorption disappeared, and then 5 mL of absolute EtOH and 2 equiv of NaBH₃CN were added. After 24 h at room temperature, the solution was made alkaline with 10% NaOH and extracted with CH₂Cl₂. Anhydrication and evaporation of the solvent gave a mixture that was purified by column chromatography using eluent B to give **49** (64% yield): IR (Nujol) ν 3200–3400 (NH) cm⁻¹; ¹H NMR (CDCl₃) δ 0.88 (t, *J* = 6.9 Hz, 3H, CH₃CH₂), 1.08–1.30 (m, 24H, 12CH₂C), 1.30–1.52 (m, 6H, CH₂CH₂N + 2CH₂ piperidine), 1.72–1.90 (m, 4H, 2CH₂ piperidine), 1.90–2.08 (m, 4H, CH₂N piperidine), 2.29 (t, *J* = 7.7 Hz, 2H, CH₂CH₂N), 2.49–2.68 (m, 2H, 2CH), 2.78–2.95 (m, 4H, 2CH₂N piperidine), 3.50 (s, 2H, CH₂Ph), 7.20–7.37 (m, 5H, CH aromatics). Anal. (C₃₂H₅₇N₃) C, H, N.

Using the same method and starting from **59** and benzylamine, compound **60** was obtained. The compound was obtained by purification of the reaction mixture by column chromatography using eluent C.

[1-Pentadecyl-(4'-piperidyl)-4-piperidylamine (22). Ten percent Pd/C (0.7 g × 1 g of benzyl derivative) was added to **49** in absolute EtOH and hydrogenated at 53 psi for 24 h at room temperature. The carbon was filtered off and the solvent removed under vacuum to obtain pure **22**: IR (Nujol) ν 3200–3380 (NH) cm⁻¹; ¹H NMR (CDCl₃) δ 0.88 (t, *J* = 6.9 Hz, 3H, CH₃CH₂), 1.10–1.31 (m, 24H, 12CH₂C), 1.31–1.56 (m, 6H, CH₂CH₂N + 2CH₂ piperidine), 1.73–1.93 (m, 4H, 2CH₂ piperidine), 1.93–2.14 (m, 3H, CH₂N piperidine + NH), 2.29 (t, *J* = 8.0 Hz, 2H, CH₂CH₂N), 2.50–2.80 (m, 4H, CH₂NH + 2CH), 2.81–2.99 (m, 2H, CH₂N piperidine), 3.04–3.19 (m, 2H, CH₂NH); ¹³C NMR (CDCl₃) δ 14.58 (q, CH₃), 23.14 (t, CH₂), 27.55 (t, CH₂), 28.13 (t, CH₂), 29.78 (t, CH₂), 30.11 (t, CH₂), 32.35 (t, CH₂), 33.35 (t, CH₂), 34.43 (t, CH₂), 45.50 (t, CH₂), 51.45 (d, CH), 53.12 (t, CH₂), 59.28 (t, CH₂). Anal. (C₂₅H₅₁N₃) C, H, N.

Using the same method, starting from **51–54**, **57**, **58**, and **60**, compounds **25**, **27**, **30**, and **32** were obtained pure, whereas **24**, **26**, and **29** were purified by column chromatography using the eluent reported in Table 2.

tert-Butyl-(1'-methyl-1'-pentadecyl-[4,4']-bipiperidinium iodide)-1-carboxylate (50). Compound **41** (60 mg) in anhydrous ether was added with an excess of CH₃I and left at room temperature for 48 h. The salt was filtered under vacuum and purified by column chromatography using eluent B: ¹H NMR (CDCl₃) δ 0.86 (t, *J* = 6.6 Hz, 3H, CH₃CH₂), 1.10–1.35 (m, 24H, 12CH₂C), 1.36–1.40 (m, 2H, CH₂CH₂N), 1.43 (s, 9H, CH₃C), 1.62–1.73 (m, 12H, CH₂ piperidine), 1.84–2.02

(m, 2H, CH₂ piperidine), 2.64 (t, *J* = 7.8 Hz, 2H, CH₂CH₂N), 3.20 (s, 3H, CH₃N), 3.53–3.61 (m, 2H, CH₂ piperidine), 4.08–4.14 (m, 2H, CH₂ piperidine). Anal. (C₃₁H₆₁LN₂O₂) C, H, N.

1-Methyl-1-pentadecyl-[4,4′]-bipiperidinium Chloride (23). Two equivalents of 6 M HCl were added to **50** dissolved in CH₃CN. After 2 h at room temperature and 5 min at 60 °C, the solution was made alkaline with 10% NaOH and extracted with CH₂Cl₂. Anhydri-fication and evaporation of the solvent gave **23** in 30% yield: mp 180–181 °C from anhydrous ether; ¹H NMR (CDCl₃) δ 0.88 (t, *J* = 6.6 Hz, 3H, CH₃CH₂), 1.19–1.31 (m, 24H, 12CH₂C), 1.32–1.39 (m, 2H, CH₂CH₂N), 1.65–1.89 (m, 12H, CH₂ piperidine), 1.91–2.02 (m, 2H, CH₂ piperidine), 2.61 (t, *J* = 7.7 Hz, 2H, CH₂CH₂N), 3.02–3.14 (m, 2H, CH₂ piperidine), 3.21 (s, 3H, CH₃N), 3.32–3.43 (m, 2H, CH₂ piperidine), 3.65 (bs, 1H, NH). Anal. (C₂₆H₅₃ClN₂) C, H, N.

(1-Benzyl-4-piperidinyl)tridecylamine (51). Two equivalents of NEt₃ and 1 equiv of bromotridecane were added to 4-amino-1-benzylpiperidine in CH₃CN. After 3 h at reflux, the solvent was removed, the residue made alkaline with 10% NaOH, and the solution extracted with CH₂Cl₂. Anhydri-fication and evaporation of the solvent gave a mixture from which **51** was purified by flash chromatography using the eluent reported in Table 6: IR (neat) ν 3300 (NH) cm⁻¹; ¹H NMR (CDCl₃) δ 0.89 (t, *J* = 6.6 Hz, 3H, CH₃CH₂), 1.21–1.37 (m, 20H, 10CH₂C), 1.37–1.57 (m, 4H, CH₂CH₂NH + CH₂ piperidine), 1.80–1.93 (m, 2H, CH₂ piperidine), 1.93–2.09 (m, 2H, CH₂N), 2.39–2.53 (m, 1H, CH), 2.62 (t, *J* = 7.3 Hz, 2H, CH₂CH₂NH), 2.80–2.93 (m, 2H, CH₂N), 3.50 (s, 2H, CH₂Ph), 7.21–7.38 (m, 5H, CH aromatics). Anal. (C₂₅H₄₄N₂) C, H, N.

Using the method described above, starting from ethyl isonipecotatate, 1-piperidone hydrochloride monohydrate, 1-nitrosopiperazine, 1-amino-4-benzylpiperazine, **30**, and **63**⁴⁶ and using the proper bromoalkane or alkanoyl chloride, compounds **48**,³⁹ **52**, **53**, **59**, **61**, and **62** were obtained, which were purified by flash chromatography using the eluent reported in Table 6. Compounds **65** and **64** were purified by column chromatography (eluent in Table 6), whereas compounds **31a**, **54**, and **55** were obtained pure.

In the same reaction of **30**, when DMF was used instead of CH₃CN, compound **31b** was obtained: IR (Nujol) ν 3300–3200 (NH), 1645 (NHCOH), 1633 (CONH) cm⁻¹; ¹H NMR (CDCl₃) δ 0.88 (t, *J* = 6.8 Hz, 3H, CH₃CH₂), 1.20–1.37 (m, 24 H, 12 CH₂C), 1.41–1.56 (m, 2H, CH₂CH₂N), 1.66–1.92 (m, 4H, 2CH₂-piperidine), 1.94–2.08 (m, 3H, CH₂N + CH), 2.42 (t, *J* = 6.3 Hz, 2H, NCH₂CH₂NH), 2.78–2.98 (m, 2H, CH₂N), 3.18–3.26 (m, 2H, CH₂NHCO), 3.28–3.41 (m, 2H, CH₂NHCHO), 5.60 (bs, 1H, NH), 6.18 (bs, 1H, NH); ¹³C NMR (CDCl₃) δ 14.48 (q, CH₃), 23.12 (t, CH₂), 27.34 (t, CH₂), 29.38 (t, CH₂), 29.73 (t, CH₂), 29.78 (t, CH₂), 30.11 (t, CH₂), 32.33 (t, CH₂), 35.04 (t, CH₂), 39.72 (t, CH₂), 39.85 (t, CH₂), 43.75 (d, CH), 53.31 (t, CH₂), 53.54 (t, CH₂), 56.80 (t, CH₂), 161.56 (d, CHO), 175.04 (s, CO). Anal. (C₂₄H₄₇N₃O₂) C, H, N.

1-Benzylpiperidine-4-carboxylic acid pentadecylamide (57). To *N*-benzylisonipecotic acid **56**⁴¹ (0.55 g; 2.1 mmol) in CHCl₃, cooled to 0 °C and under nitrogen was added 0.39 mL (5.4 mmol) of NEt₃ and 3.1 mL (3.2 mmol) of ethyl chloroformate. After 1 h, pentadecylamine (0.73 g; 3.2 mmol) in a few milliliters of CHCl₃ was added, and the mixture was kept at room temperature for 20 min. The solution was then basified with NaHCO₃ and extracted with CH₂Cl₂. Anhydri-fication and evaporation of the solvent gave a residue that was purified by chromatography using eluent A, to give **57** (yield 56%): mp 47–50 °C; IR (Nujol) ν 3307 (NHCO), 1640 (CONH) cm⁻¹; ¹H NMR (CDCl₃) δ 0.88 (t, *J* = 6.6 Hz, 3H, CH₃CH₂), 1.17–1.40 (m, 24H, 12CH₂C), 1.40–1.59 (m, 2H, CH₂CH₂NH), 1.67–1.91 (m, 4H, 2CH₂ piperidine), 1.91–2.18 (m, 3H, CH₂N + CH), 2.86–3.00 (m, 2H, CH₂N), 3.23 (m, 2H, CH₂CH₂NH), 3.49 (s, 2H, CH₂Ph), 5.57 (bt, 1H, NH), 7.20–7.39 (m, 5H, CH aromatics). Anal. (C₂₈H₄₈N₂O) C, H, N.

[1-(2-Aminoethyl)-4-piperidylmethyl]pentadecylamine (33). Borane methyl sulfide complex (2 M in THF; 2.2 mmol) was added to **31a** (0.75 mmol) in THF anhydrous at reflux under nitrogen. After 1 h, the solution was treated with 4.9 mmol of 6 M HCl and maintained at reflux for 30 min.

The cold solution was made alkaline with 10% NaOH and extracted with CH₂Cl₂. Anhydri-fication and evaporation of the solvent gave a mixture that was purified by column chromatography with the eluent reported in Table 3 to give **33**: ¹H NMR (CDCl₃) δ 0.88 (t, *J* = 6.6 Hz, 3H, CH₃CH₂), 1.18–1.32 (m, 24H, CH₂-C), 1.35 (bs, 2H, NH₂), 1.43–1.55 (m, 2H, CH₂-CH₂-NH₂), 1.63–1.74 (m, 3H, CH₂ piperidine + CH), 1.86–2.02 (m, 2H, CH₂ piperidine), 2.38 (t, *J* = 6.1 Hz, 2H, CHCH₂NH₂), 2.47 (d, *J* = 4.0 Hz, 2H, NCH₂CH), 2.50–2.61 (m, 2H, CH₂ piperidine), 2.78 (t, *J* = 6.1 Hz, 2H, CH₂CH₂NH₂), 2.83–2.95 (m, 4H, CH₂ piperidine + NH₂CH₂C); ¹³C NMR (CDCl₃) δ 17.69 (q), 23.12 (t), 27.78 (t), 29.78 (t), 30.04 (t), 30.11 (t), 31.04 (t), 32.35 (t), 36.35 (d), 39.38 (t), 50.54 (t), 54.22 (t), 56.18 (t), 61.63 (t). Anal. (C₂₃H₄₉N₃) C, H, N.

Using the same method and starting from **57**, compound **58** was obtained and purified by flash chromatography.

1-Pentadecyl-4-piperidinylamine (28). To 1 equiv of **48** in anhydrous MeOH were added molecular sieves (3 Å) (1 g of derivative × 8 g of molecular sieves), 10 equiv of ammonium acetate, and 1 equiv of NaBH₃CN, under N₂. After 24 h at room temperature, the solvent was removed, 5 mL of H₂O was added, and the solution was adjusted at pH 2 with 6 N HCl. After evolution of gas had ceased, the solution was neutralized with KOH, the salts were filtered off, and the solution was extracted with CH₂Cl₂. Anhydri-fication and evaporation of the solvent gave **28** pure: IR (Nujol) ν 3300–3400 (NH₂) cm⁻¹; ¹H NMR (CDCl₃) δ 0.86 (t, *J* = 6.9 Hz, 3H, CH₃CH₂), 1.15–1.39 (m, 26H, 12CH₂C + CH₂ piperidine), 1.39–1.55 (m, 2H, CH₂CH₂N), 1.73–1.89 (m, 2H, CH₂ piperidine), 1.89–2.07 (m, 2H, CH₂N piperidine), 2.08 (bs, 2H, NH₂), 2.29 (t, *J* = 8.0 Hz, 2H, CH₂CH₂N), 2.56–2.73 (m, 1H, CH), 2.79–2.94 (m, 2H, CH₂N piperidine); ¹³C NMR (CDCl₃) δ 14.56 (q, CH₃), 23.12 (t, CH₂), 27.58 (t, CH₂), 28.15 (t, CH₂), 29.78 (t, CH₂), 30.02 (t, CH₂), 30.09 (t, CH₂), 32.33 (t, CH₂), 36.32 (t, CH₂), 49.19 (d, CH), 53.00 (t, CH₂), 59.28 (t, CH₂). Anal. (C₂₀H₄₂N₂) C, H, N.

Using the same method, starting from **59**, compound **29** was obtained.

4-Pentadecylpiperazin-1-ylamine (34). Compound **61** (0.58 mmol) in 0.6 mL of H₂O and 0.3 mL of MeOH cooled at 0 °C was added to 0.7 mmol of concentrated HCl and 1.5 mmol of Zn. After a while, 3.2 mmol of concentrated HCl was added, keeping the mixture for 30 min at room temperature. Finally, the suspension was refluxed for 2 h. The mixture was then basified with 50% NaOH and extracted with CH₂Cl₂. Anhydri-fication and evaporation of the solvent gave a mixture that was purified by column chromatography with the eluent reported in Table 4 to give **34**: IR (Nujol) ν 3300–3600 (NH₂) cm⁻¹; ¹H NMR (CDCl₃) δ 0.89 (t, *J* = 6.9 Hz, 3H, CH₃CH₂), 1.16–1.38 (m, 24H, 12CH₂C), 1.40–1.64 (m, 2H, CH₂CH₂N), 1.58 (bs, 2H, NH₂), 2.31 (t, *J* = 8.0 Hz, 2H, CH₂CH₂N), 2.36–2.48 (m, 4H, 2CH₂ piperazine), 2.91 (t, *J* = 5.1 Hz, 4H, 2CH₂ piperazine); ¹³C NMR (CDCl₃) δ 14.56 (q, CH₃), 23.12 (t, CH₂), 27.11 (t, CH₂), 28.05 (t, CH₂), 29.78 (t, CH₂), 30.02 (t, CH₂), 30.09 (t, CH₂), 32.33 (t, CH₂), 46.45 (t, CH₂), 55.02 (t, CH₂), 59.90 (t, CH₂). Anal. (C₁₉H₄₁N₃) C, H, N.

With the same method, starting from **62**, compound **35** was obtained. Both compounds were transformed into the hydrochlorides as described before.

Pentadecylpiperazin-1-ylamine (36). Five equivalents of HCOONH₄ and 0.5 equiv of 10% Pd/C were added to a solution of **64** in anhydrous MeOH under N₂. The mixture was refluxed for 8 h, then the carbon was filtered off and the solution was evaporated. The residue was made alkaline with 10% NaOH and extracted with CH₂Cl₂. Anhydri-fication and evaporation of the solvent afforded **36** pure: IR (Nujol) ν 3200–3400 (NH) cm⁻¹; ¹H NMR (CDCl₃) δ 0.88 (t, *J* = 6.6 Hz, 3H, CH₃CH₂), 1.13–1.38 (m, 24H, 12CH₂C), 1.40–1.63 (m, 3H, CH₂CH₂N + NH), 2.30 (t, *J* = 7.9 Hz, 2H, CH₂CH₂N), 2.36–2.49 (m, 4H, 2CH₂NH piperazine), 2.89 (t, *J* = 5.1 Hz, 4H, 2CH₂NNH); ¹³C NMR (CDCl₃) δ 14.56 (q, CH₃), 23.12 (t, CH₂), 27.09 (t, CH₂), 28.07 (t, CH₂), 29.78 (t, CH₂), 30.02 (t, CH₂), 30.09 (t, CH₂), 32.33 (t, CH₂), 46.48 (t, CH₂), 55.04 (t, CH₂), 59.92 (t, CH₂). Anal. (C₁₉H₄₁N₃) C, H, N.

With the same method, starting from **65**, pure **37** was obtained.

Experimental Pharmacology

Isolation of Human Lymphocytes. Peripheral blood samples (15–20 mL) were immediately treated with heparin sodium (500 IU of heparin/10 mL of blood) to prevent coagulation and diluted 1:1 with saline solution. The lymphocytes were then isolated by subjecting the diluted blood to Ficoll-isopaque (istopaque) density-gradient centrifugation as described by Böyum.⁵⁶ Mononuclear cells were counted.

Adenylyl Cyclase Activity Assay in Intact Lymphocytes. The activation of Gi proteins was assessed by evaluating the ability of the investigated compounds to inhibit stimulated adenylyl cyclase activity in peripheral blood intact lymphocytes. Lymphocytes from healthy subjects were preincubated with or without pertussis toxin (PTX) at the concentration of 100 ng/mL at 37 °C for 90 min as previously described.⁵⁰ For the determination of adenosine 3',5'-cyclic monophosphate (cAMP) content, lymphocytes [(1–2) × 10⁶ intact cells/assay] were incubated either with dimethyl sulfoxide (DMSO) or 10⁻⁴ M forskolin, which stimulated the cAMP formation.⁵⁰ To obtain the inhibition curve of cAMP production, the nonhydrolyzable analogue of GTP, Gpp(NH)p (guanylyl 5'-imidophosphate), or the investigated molecules were added to the forskolin-containing samples in concentrations ranging from 1 nM to 100 μM, in a final volume of 300 μL. After a 15-min incubation at 37 °C, the cells were lysed and the samples centrifuged according to the method of Brodde et al.⁵⁷ The cAMP content was determined in a 100-μL aliquot of the supernatant using an enzyme immunoassay kit (Amersham). Protein concentration was determined by the bicinchoninic acid protein assay kit (Sigma) with bovine serum albumin as standard. All assays were performed in duplicate. The recovery of a known amount of cAMP (10 pmol), added to the incubation mixture after boiling, was found to be 94.2 ± 7.8% (n = 6). For compounds **1**, **5**, **27**, **28**, and **31a** the inhibition of the adenylyl cyclase activity was also detected on lymphocytes that were first permeabilized with the detergent digitonin (10 μg/mL) under such conditions to maintain the biological functions intact.⁵⁸

Expression and Purification of G Proteins. The CDS (coding sequence) encoding for α inhibitory subunit isoforms 1 ($\alpha_{i/1}$) and 3 ($\alpha_{i/3}$) and for α other subunit isoform 1 ($\alpha_{o/1}$) were obtained by RT-PCR technique. Total RNA from human brain and heart (brain for isoforms $\alpha_{i/1}$ and $\alpha_{o/1}$, heart for isoform $\alpha_{i/3}$) was extracted by TriReagent (Sigma) according to the manufacturer's protocol. Five hundred nanograms of total RNA was reverse transcribed and amplified by the SuperScript One-Step RT-PCR system (Invitrogen) (primer sequences were as follows):



PCRs were performed for 35 cycles using 15 s of denaturation at 94 °C, 30 s of annealing at 60 °C, and 1 min/kilobase extension at 68 °C.

All PCR products were cloned into pCR-TOPO T7 expression vector and transformed into competent *Escherichia coli* BL21/DE3. Protein expression was induced by adding 1 mM isopropyl- β -D-thiogalactopyranoside (IPTG) and 34 μg/mL chloramphenicol. Bacteria were collected by centrifugation at 4 °C for 15 min at 5000 rpm, and cell pellets were resuspended in binding buffer (50 mM Hepes, pH 8.0). The solution was then sonicated on ice with six 10-s bursts at high intensity with a

10-s cooling period between each burst. After centrifugation at 3000g for 15 min to precipitate the cellular debris, the supernatant was transferred to a fresh tube and 5 μL of the lysate was removed for SDS-PAGE analysis. Purification was performed by immobilized metal affinity chromatography (IMAC) exploiting the 6xHistidine-tag fused to the recombinant protein.

GTP γ S Binding. Binding of GTP γ S (the nonhydrolyzable GTP analogue) to recombinant G protein was measured according to a standard method.²² Three picomoles of α subunit in native conditions or reconstituted into lipid vesicles was incubated at 30 °C (20 °C for G α_o) for 30 min in 100 μL of a reaction mixture containing 50 mM Hepes-NaOH, 1 mM EDTA, 1 mM DTT, 100 μM MgSO₄, and 0.1 μM [³⁵S]GTP γ S (0.1 μCi/tube). Reaction was stopped by the addition of 1 mL of an ice-cold stop buffer (consisting of 10 mM Tris-HCl, pH 8.0, 25 mM MgCl₂, and 100 mM NaCl). The diluted samples were filtered through cellulose nitrate membranes (0.45-μm pore size) under weak vacuum. The filters were washed with 12 mL of the same buffer and dried. The retained radioactivity was quantified by scintillation spectroscopy. Specific binding was calculated by subtracting the GTP γ S bound in the presence of 100 μM cold GTP γ S from total binding.

Supporting Information Available: IR and ¹H NMR data (chemical shifts, ppm) and shift assignment of compounds **7–65**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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